

Operation of mixing



University of Pécs

Institute of Pharmaceutical Technology and Biopharmacy

Operation of Mixing

Definition of mixing

Mixing may be defined as a **unit operation** that aims to **homogenize two or more components**.

During mixing two or more substances get into **homogeneous distribution** in each other.

Theory of Mixing

Theory of Mixing

In pharmaceutical practice can be:

- **Self-mixing** materials (mixing occurs due to heat movements of molecules, generally a slow process. Such materials are gases and liquids with low viscosity.)
- **Non-self-mixing** materials :
 1. *Structures retaining mixture state*
 - powders,
 - liquid with high viscosity,
 - stable disperse structures.
 2. *Structures non-retaining mixture state*
 - suspension,
 - emulsion, which are separating into phaseses.

Theory of Mixing

Aims

- homogenization,
- dispersion (emulsifying, suspending),
- heating,
- cooling,
- wetting,
- drying,
- crystallization,
- disintegration,
- granulation,
- preparation of ointment and suppositories,
- microencapsulation,
- preparation of micropellets,
- preparation of nano medicines,
- chemical reaction,
- biochemical reaction, fermentation
- biopharmacy examination (e.g. dissolution, membrane permeability),
- coating.

Theory of Mixing

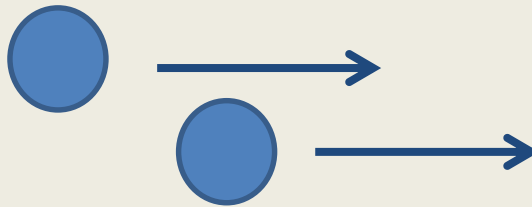
During the operation of mixing, several processes, phenomena can be regarded:

- *increase in particle size* (e.g. at granulation),
- *decrease in particle size* (e.g. at emulsifying),
- *deformation* (e.g. at disintegration),
- *flowing property of material*.

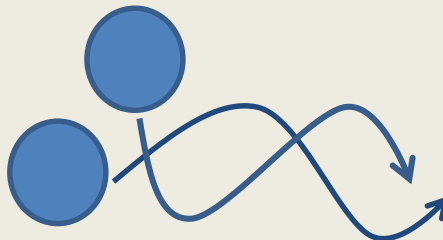
Theory of Mixing

Flow of material can be laminar and turbulent:

- At ***laminar flow***, vector of speed of particles is parallel along flow line (parallel with axle tube).
Particles move orderly next to each other without any mixing.



- At ***turbulent flow***, movement of particles shows only overall the flow line. Due to the swirling, whirling movement of the particles, the layers next to each other are blended.



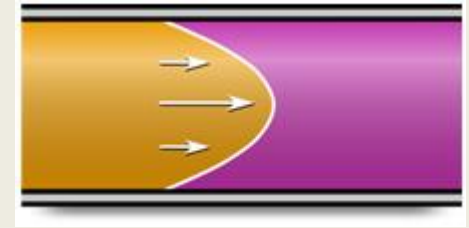
Theory of Mixing

Flow of material can be laminar and turbulent:

Laminar flow or streamline flow in pipes (or tubes) occurs when a fluid flows **in parallel layers, with no disruption** between the layers.

At low velocities, the fluid tends to flow without lateral mixing. There are no cross-currents perpendicular to the direction of flow, nor eddies or swirls of fluids.

In laminar flow, the motion of the particles of the fluid is very orderly with all particles moving in straight lines parallel to the pipe walls.



Turbulent flow is a flow regime characterized by **chaotic property changes**. This includes rapid variation of pressure and flow velocity in space and time.

In contrast to laminar flow the fluid no longer travels in layers and mixing across the tube is highly efficient.



Theory of Mixing

Euler number (Eu) is a specific non-dimensional number to flow occurring at mixing:

$$Eu = \frac{P}{d^5 n^3 \rho}$$

P = performance of stirrer

d = diameter of stirrer

n = rotational speed of stirrer

ρ = density of mixed material

Theory of Mixing

Reynolds number (Re) is also a specific non-dimensional number for mixing.

The value of Re in laminar interval is from 10 to 60, in turbulent interval is $>10^3$

$$Re = \frac{d^2 n \rho}{\mu}$$

μ = dynamic viscosity of the fluid

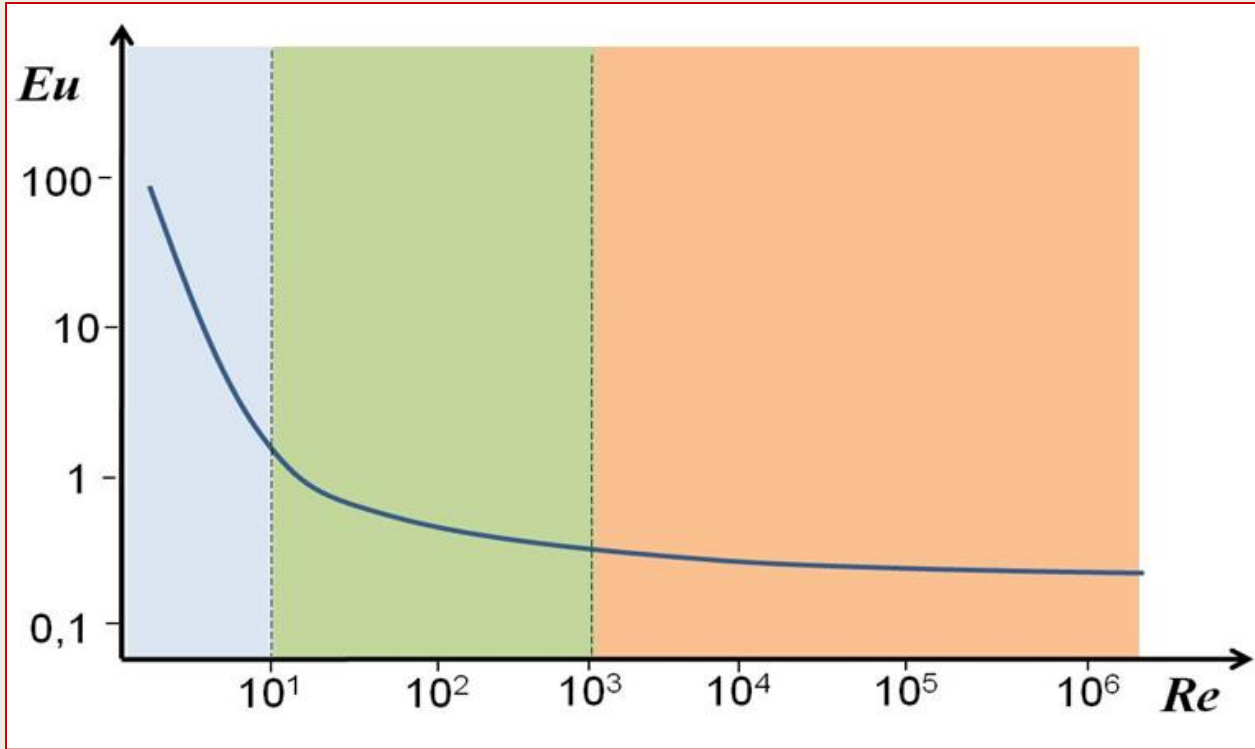
d = diameter of stirrer

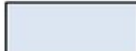


n = rotational speed of stirrer

ρ = density of mixed material

Theory of Mixing

Euler-Reynolds diagram



-  laminar
-  transitional
-  turbulent

Theory of Mixing

The **performance** required for mixing can be characterized by the following expression:

$$P = N_e \cdot \rho \cdot n^3 \cdot D^5$$

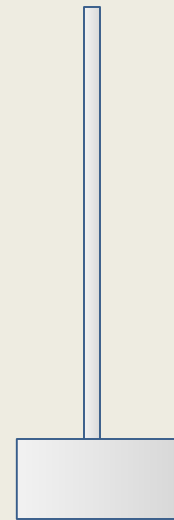
N_e = Newton-number (resistance factor)

[number without dimension]

ρ = density of material

n = speed of stirrer

D = diameter of stirrer



D

Theory of Mixing

Degree of mixing (M) can be calculated from relative standard deviation and can be characterized by the following **first-order kinetic formula**:

$$M = RSD_{\infty} + (RSD_o - RSD_{\infty})e^{-kt}$$

RSD_o = relative standard deviation in initial stage

RSD_{∞} = relative standard deviation at the measured degree of mixing

k = rate constant

t = time

Practice of mixing

Theory of Mixing

What type of mixer should be chosen?

What should be stirred?

- liquid's mixture,
 - dissolution (dissolution of solid substances),
 - suspension,
 - emulsion,
- liquid
-
- ointment,
 - paste
- semi-solid
-
- powder
 - granules
 - dragée formation
- solid

Theory of Mixing

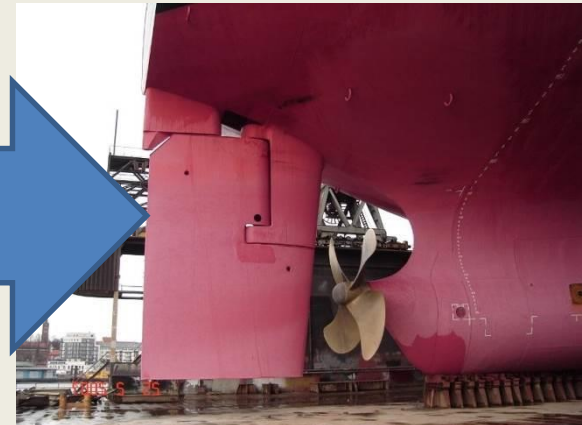
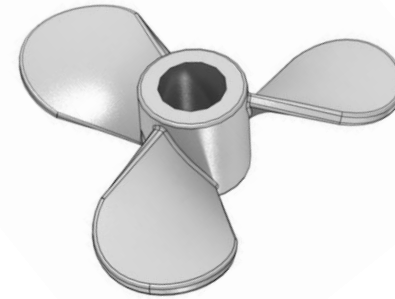
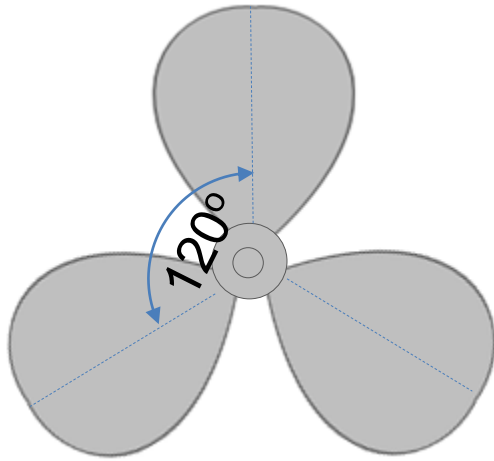
Speed of mixing:

- **slow** stirring speed **< 100 rpm**
to avoid sedimentation, for crystallization
- **medium** stirring speed **100 - 1000 rpm**
for viscous substances, syrups, ointments
- **fast** stirring speed **> 1000 rpm**
*to dissolve of solid substances,
for preparation of liquid mixtures*

Mixing of liquids

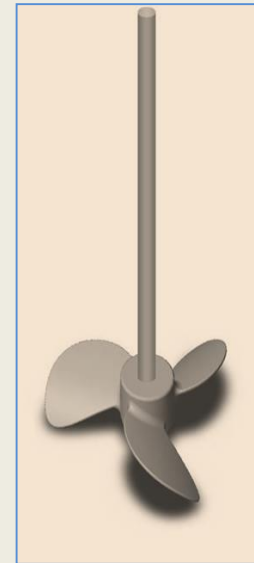
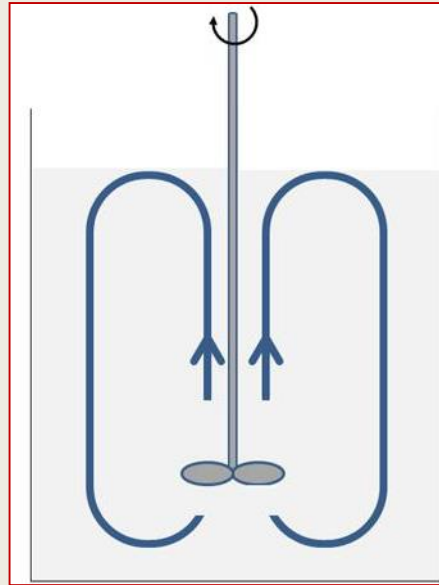
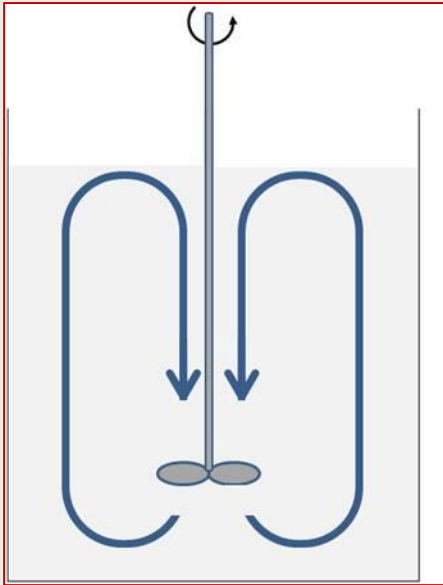
Mixing of liquids

Propellers

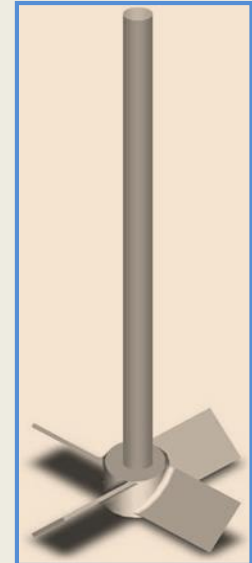


Mixing of liquids

Axial flow



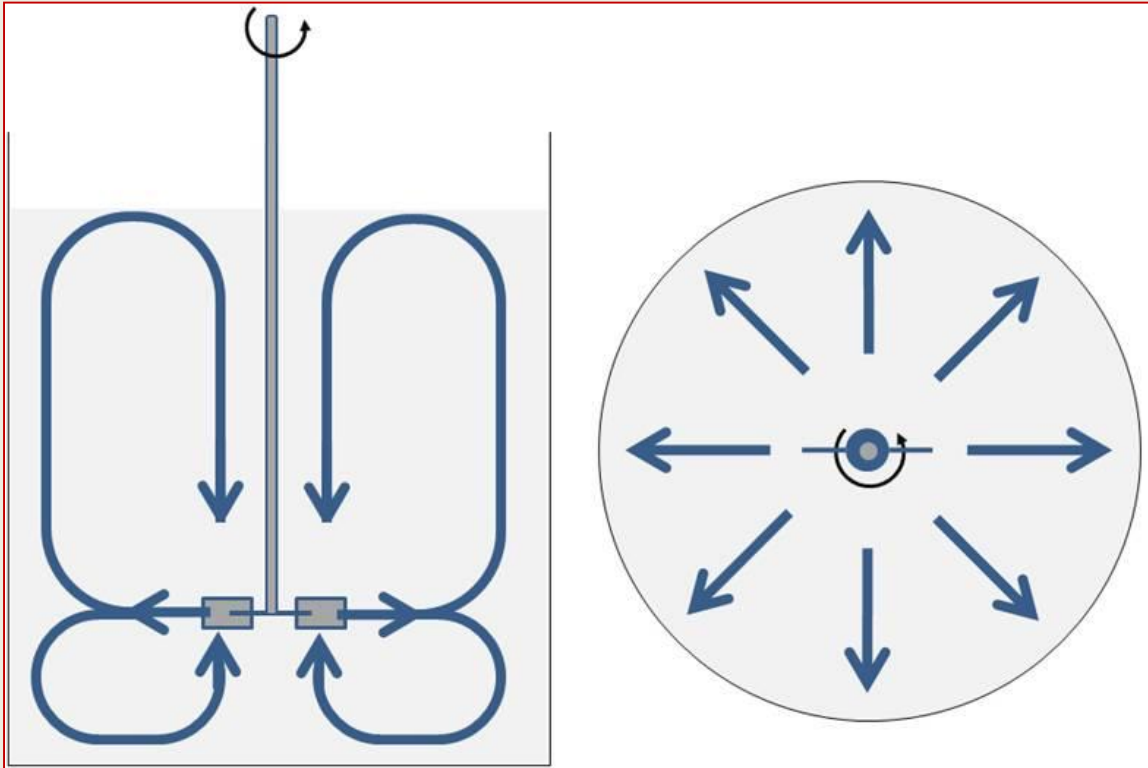
Tree blade
popeller stirrer



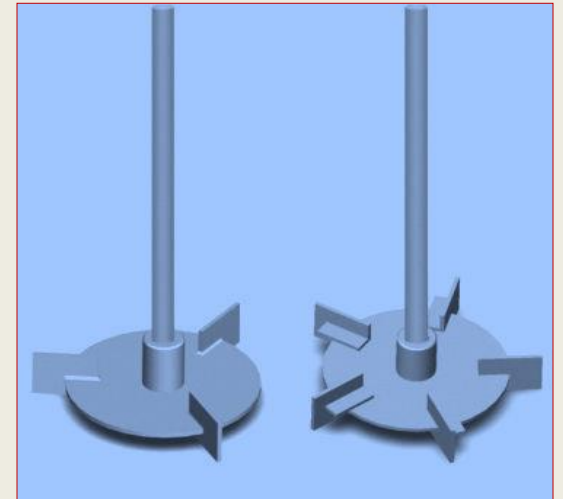
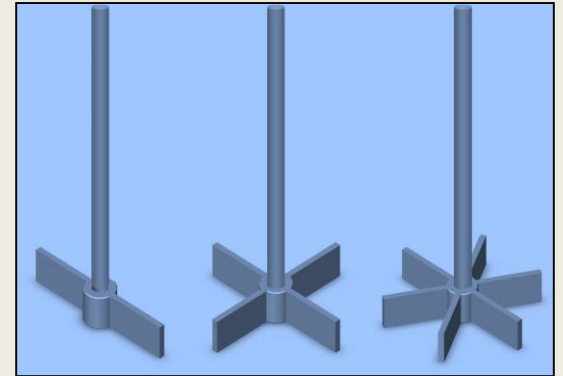
Four blade
oblique stirrer

Mixing of liquids

Radial flow



Opened turbine stirrer



Disc turbine stirrer

Mixing of liquids

Tangential flow

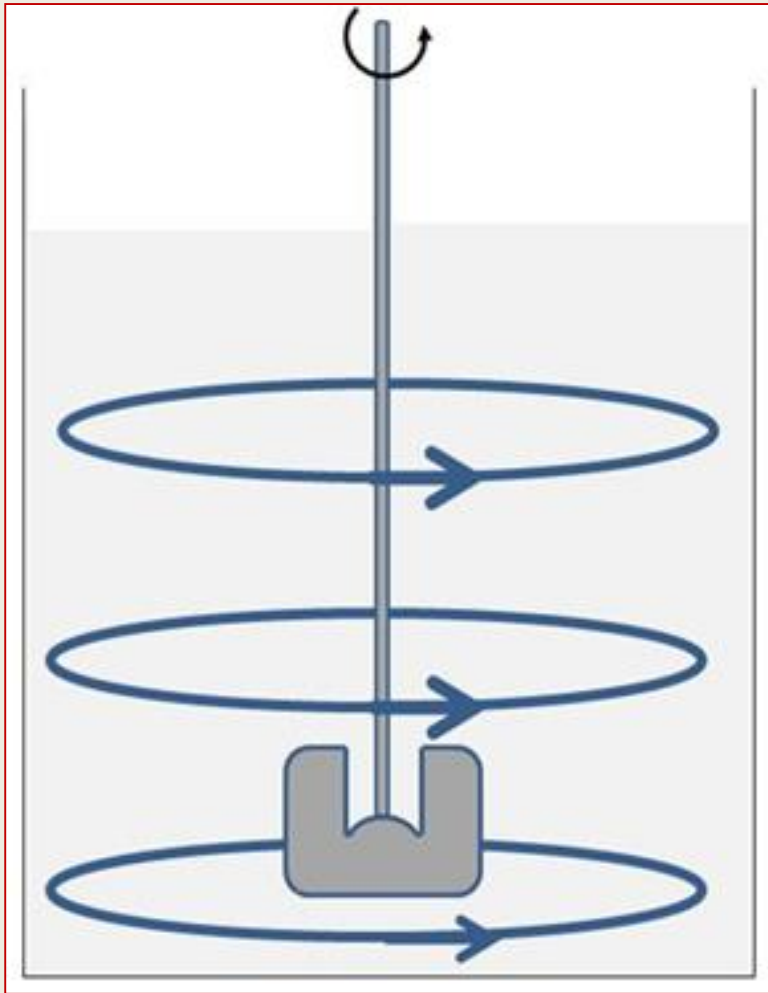


Plate stirrer

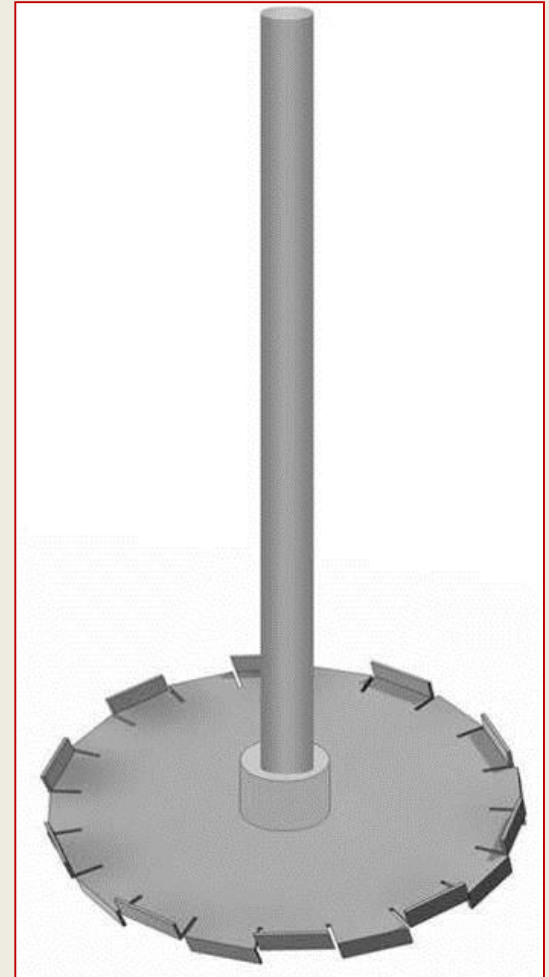
Mixing of liquids

Tangential flow



Mixing of liquids

Blade mixer



Mixing of liquids

Anchor mixer

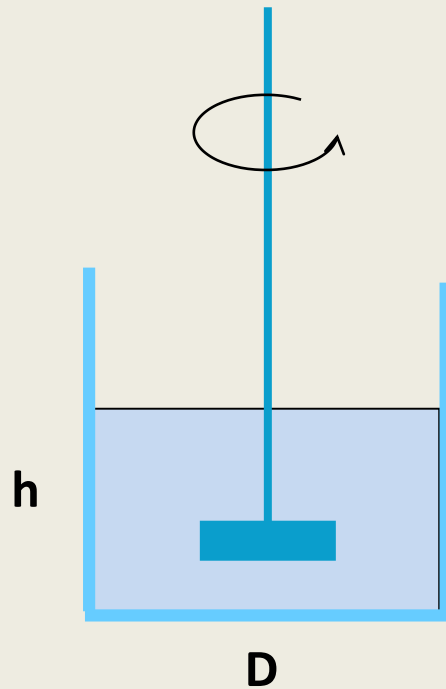


Mixing of liquids

Kneading stirrer



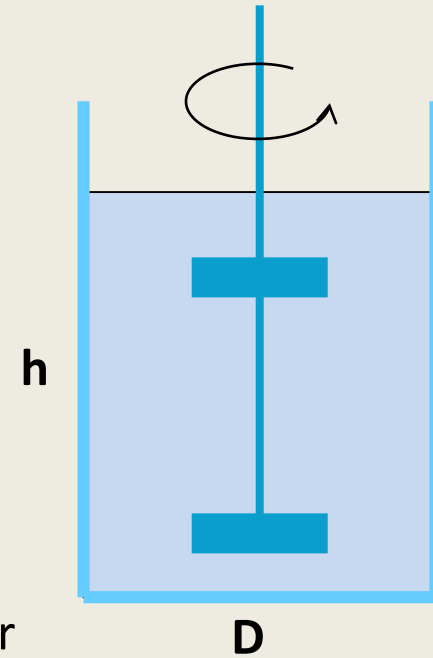
Mixing of liquids



**good wettable
materials**

$$k = \frac{h}{D} \leq 1$$

h = height
D = stirrer diameter

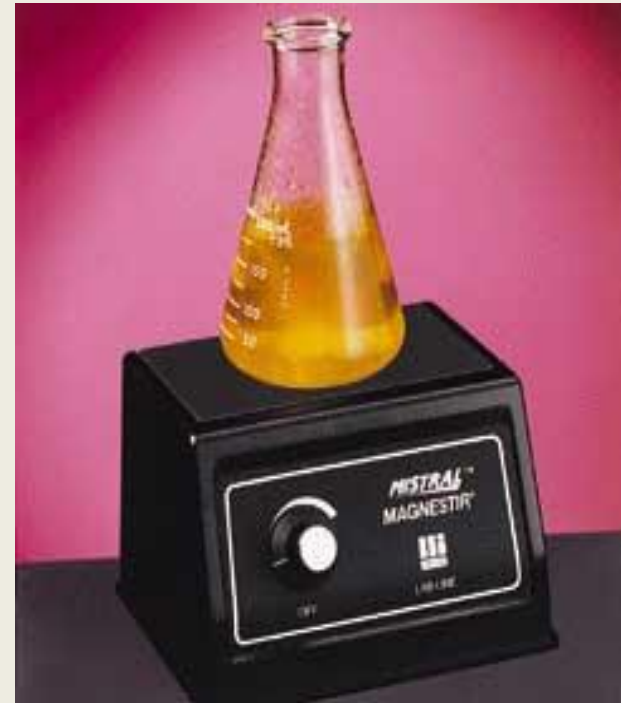


**poor wettable
materials**

$$k = \frac{h}{D} \geq 1,5$$

Mixing of liquids

magnetic stirrer



Mixing of liquids

Vortex stirrer



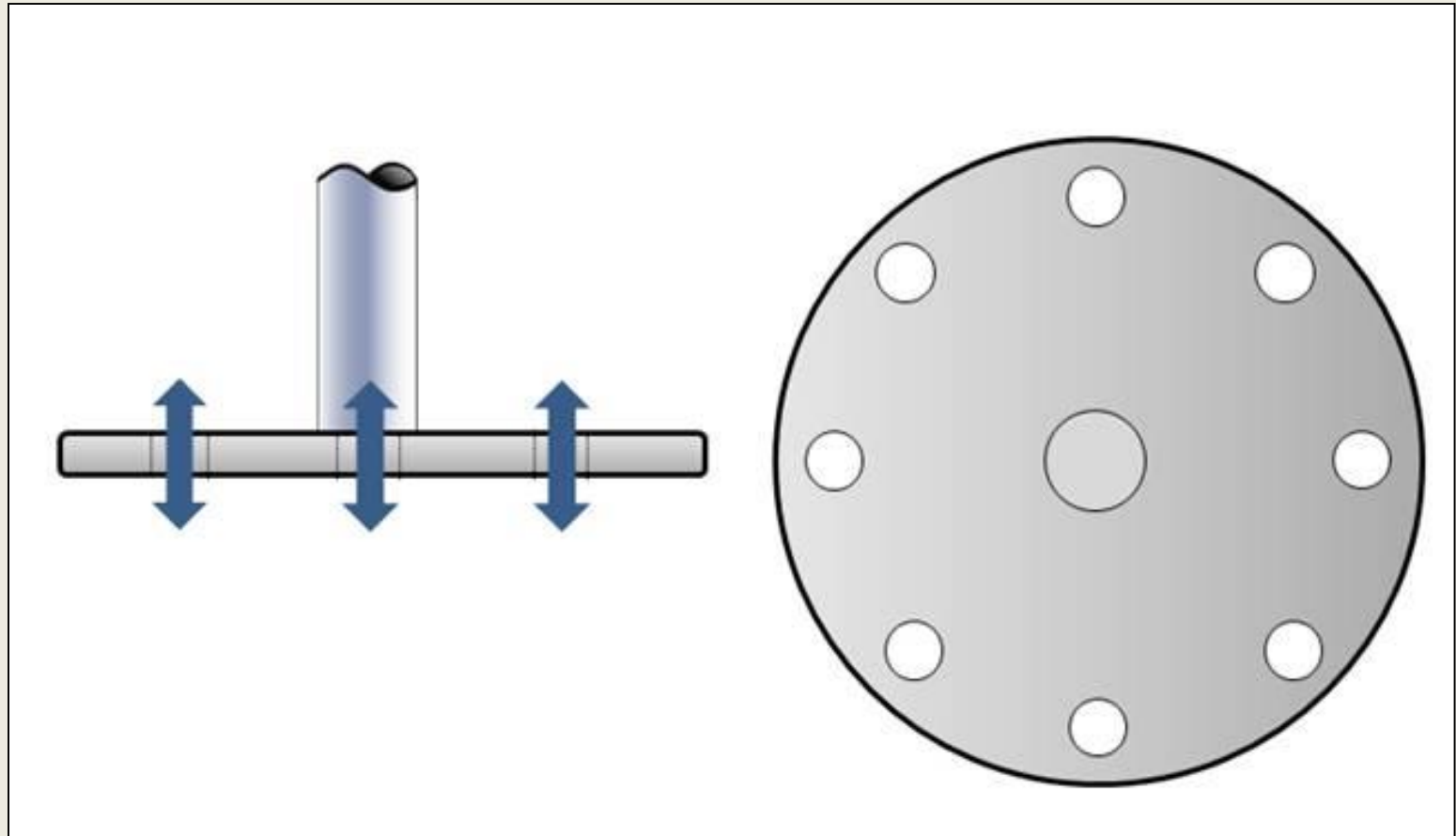
Mixing of liquids

Ultrasonic stirrer



Mixing of liquids

Vibration stirrer



Mixing of liquids

Rapid mixer



Mixing of liquids

shaker



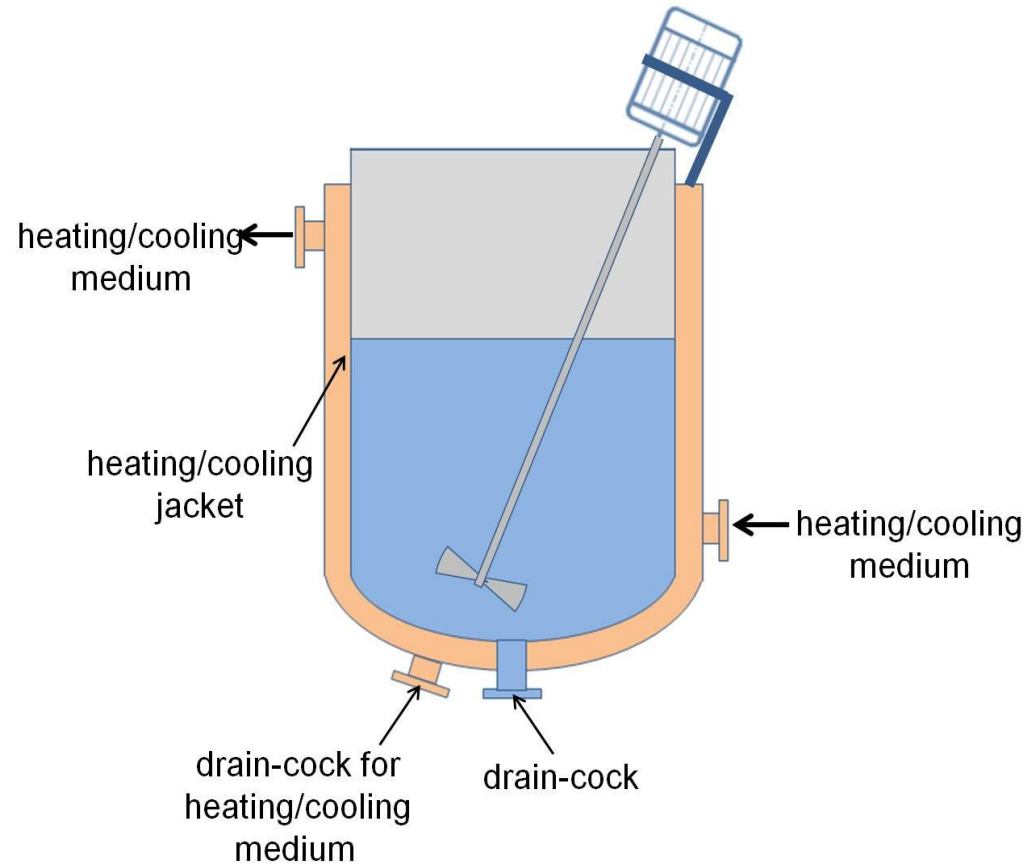
Mixing of liquids

shaker



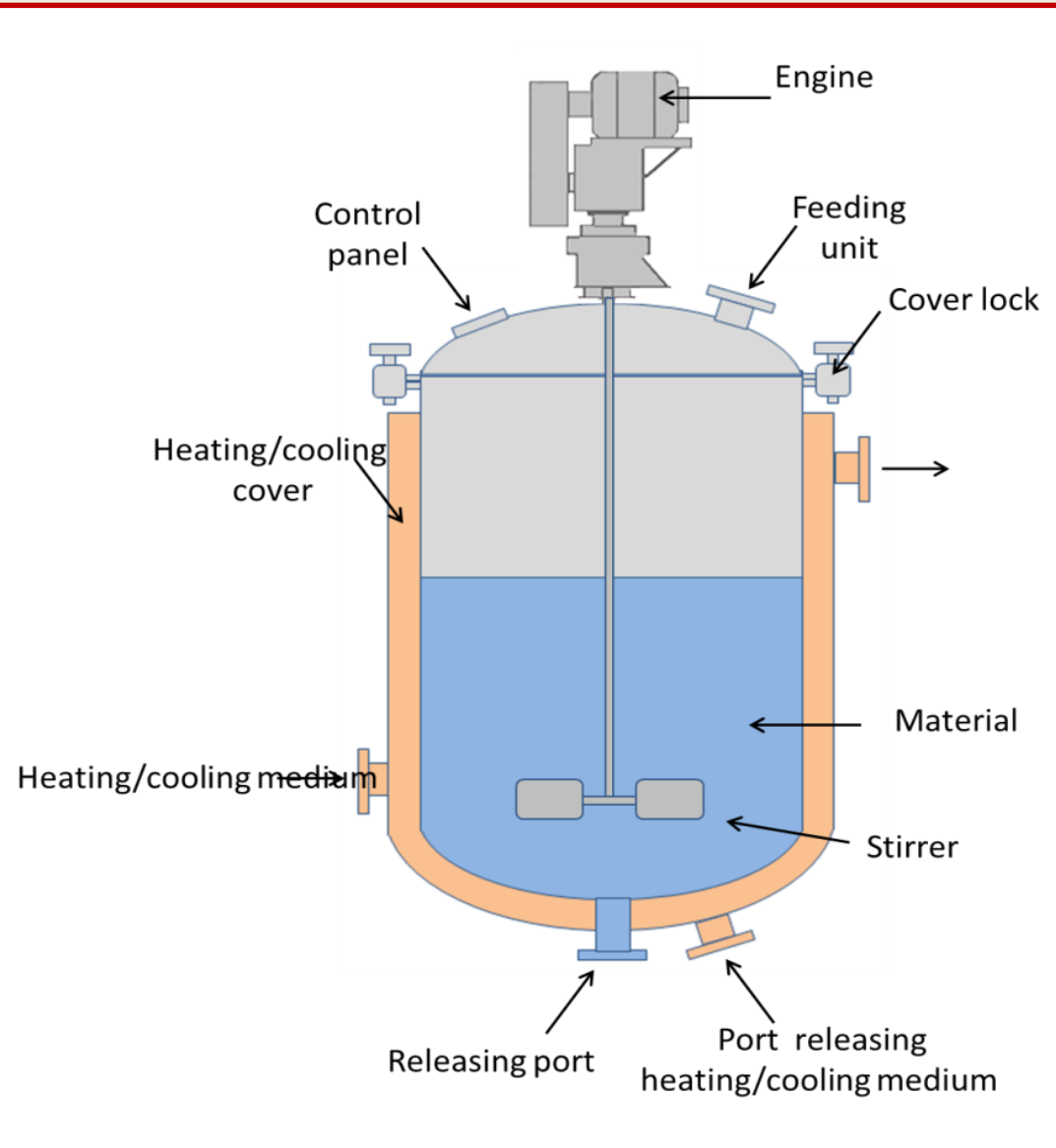
Mixing of liquids

industrial
opened duplicator with
oblique stirrer



Mixing of liquids

industrial
closed duplicator with
stirrer



Mixing of liquids

industrial closed
duplicator with stirrer



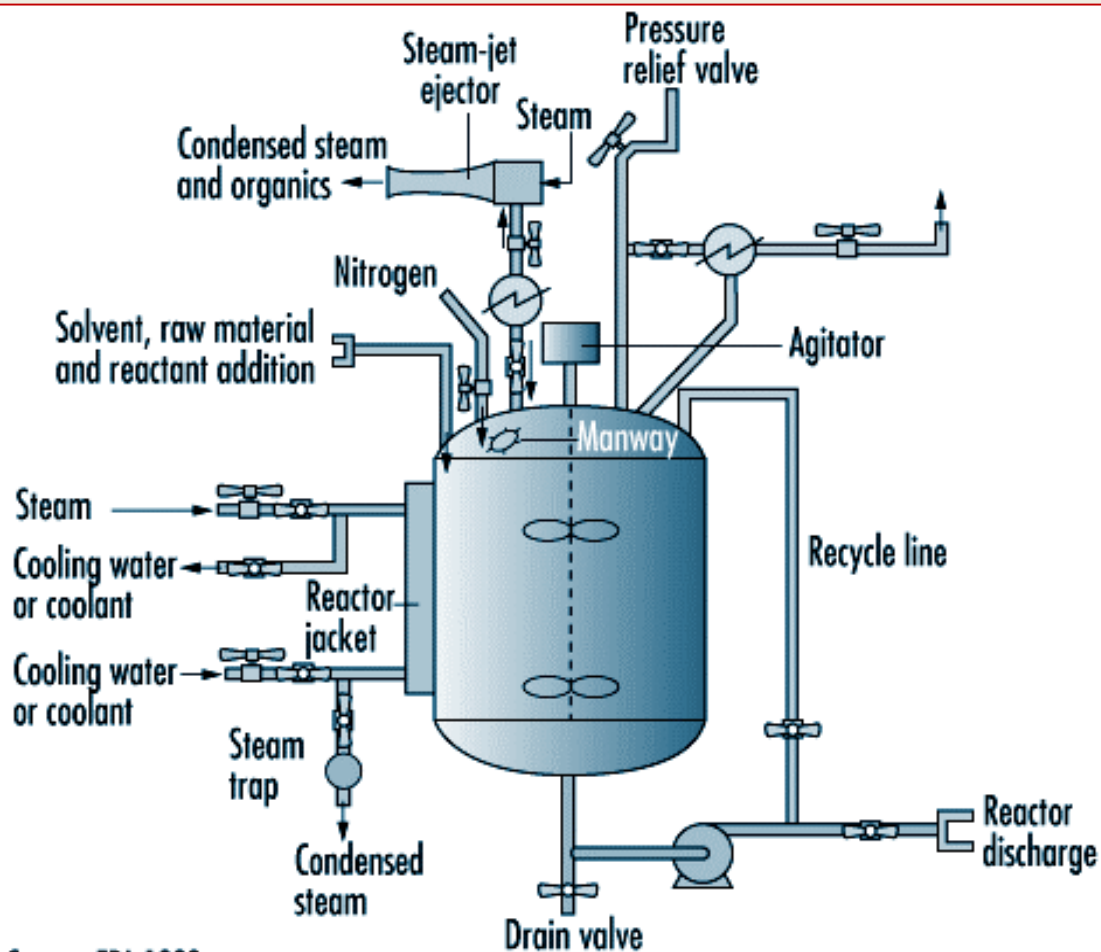
Upper and lower powered industrial
duplicators with oblique stirrer



Duplicator apparatus
with propeller stirrer

Mixing of liquids

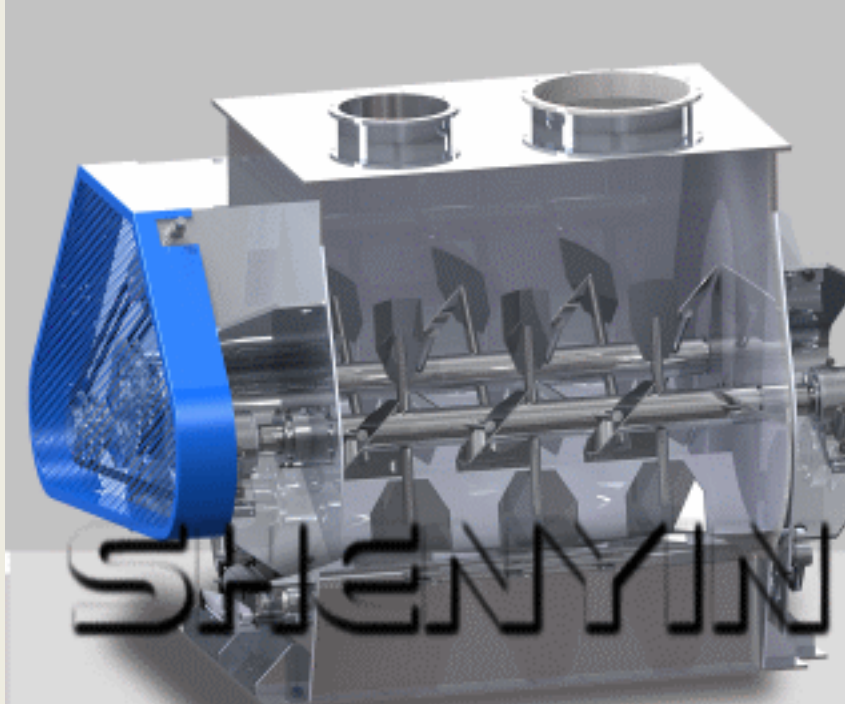
Reactors for chemical (organic synthesis), fermentors, distillation columns, crystallization and equipments for biotechnological operations



Source: EPA 1993.

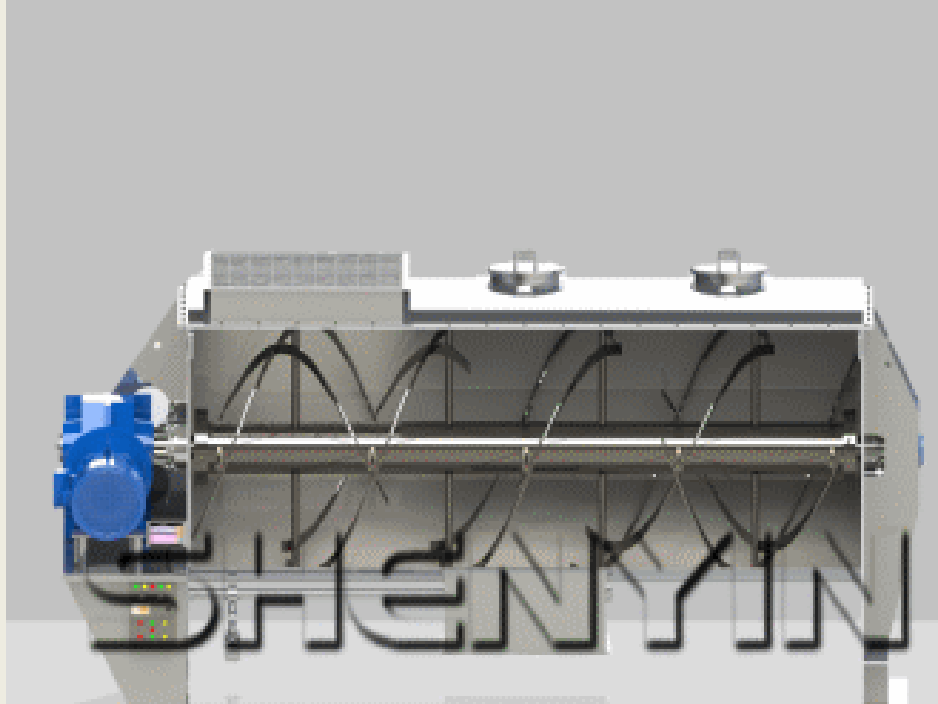
Mixing of liquids

Double Shafts Paddle Mixer



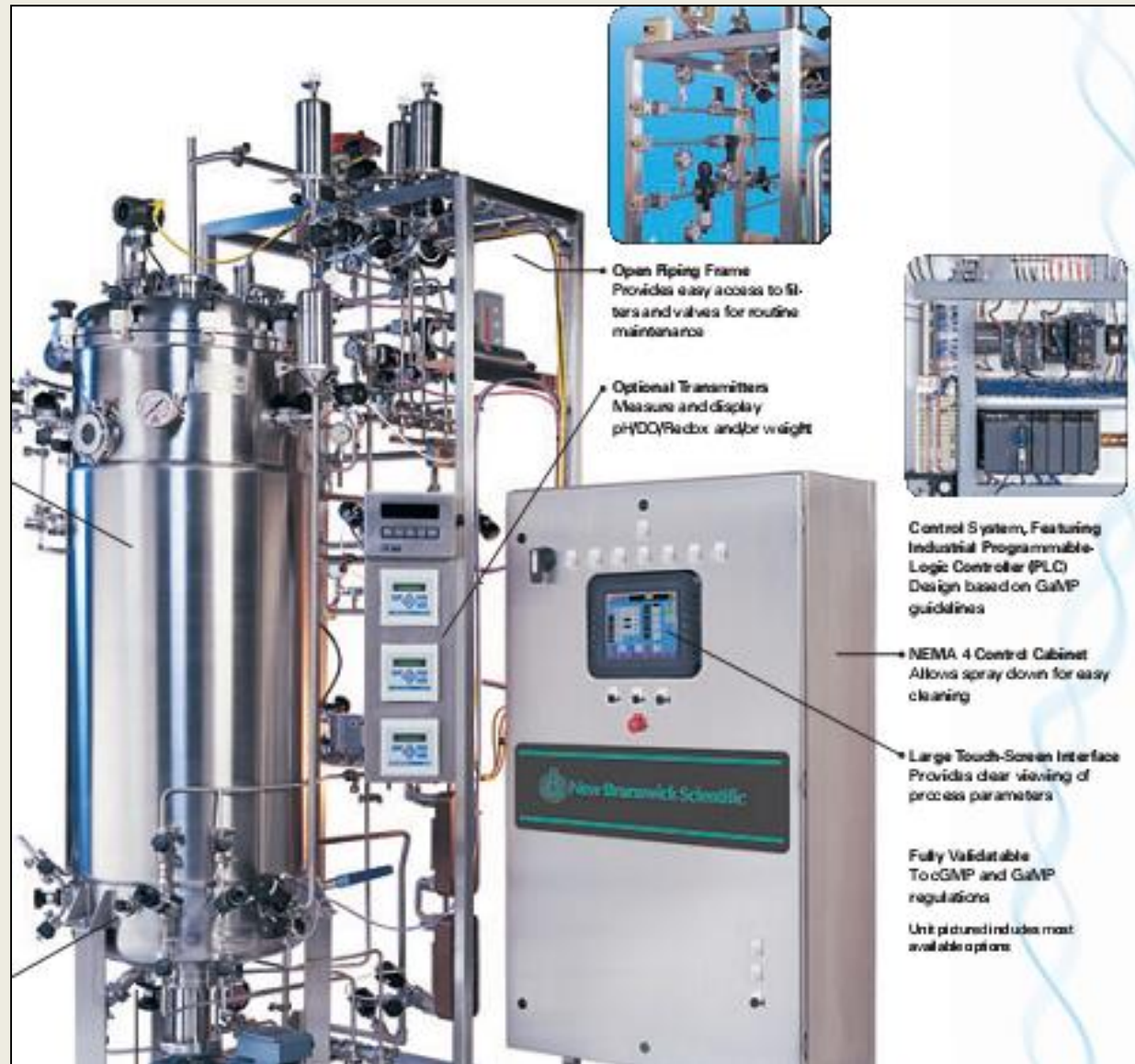
Mixing of liquids

Horizontal ribbon Mixer



Mixing of liquids

Fermenters



Mixing of liquids

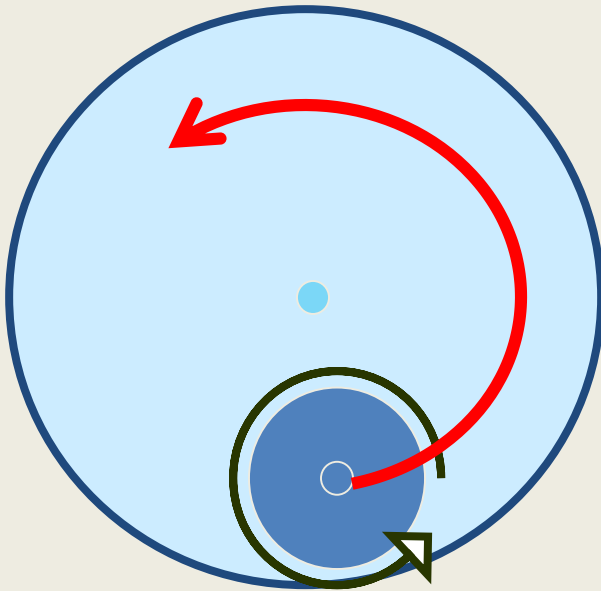
Biotechnology



Mixing of semisolids

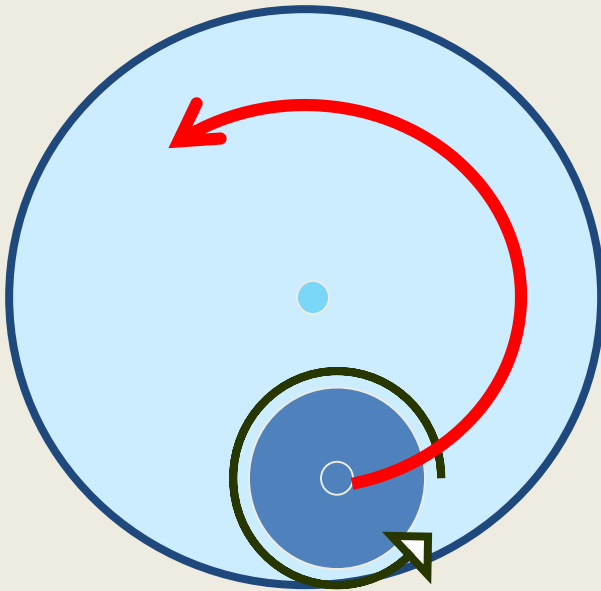
Mixing of semisolids

Planetary mixer



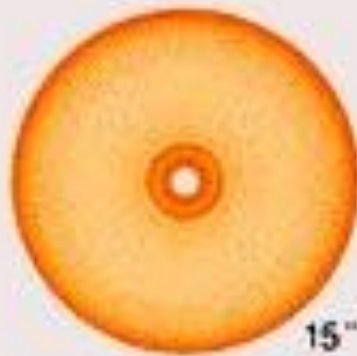
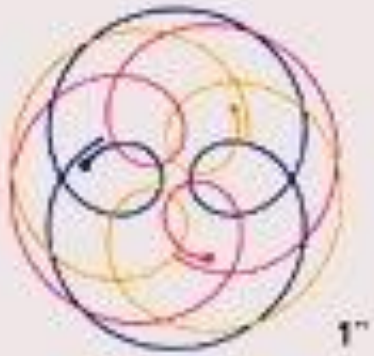
Mixing of semisolids

Planetary mixer



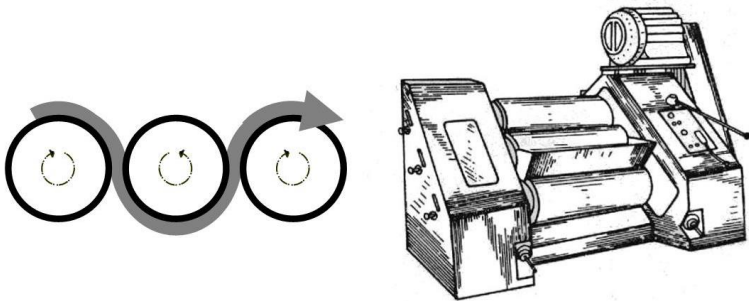
Mixing of semisolids

Planetary mixer



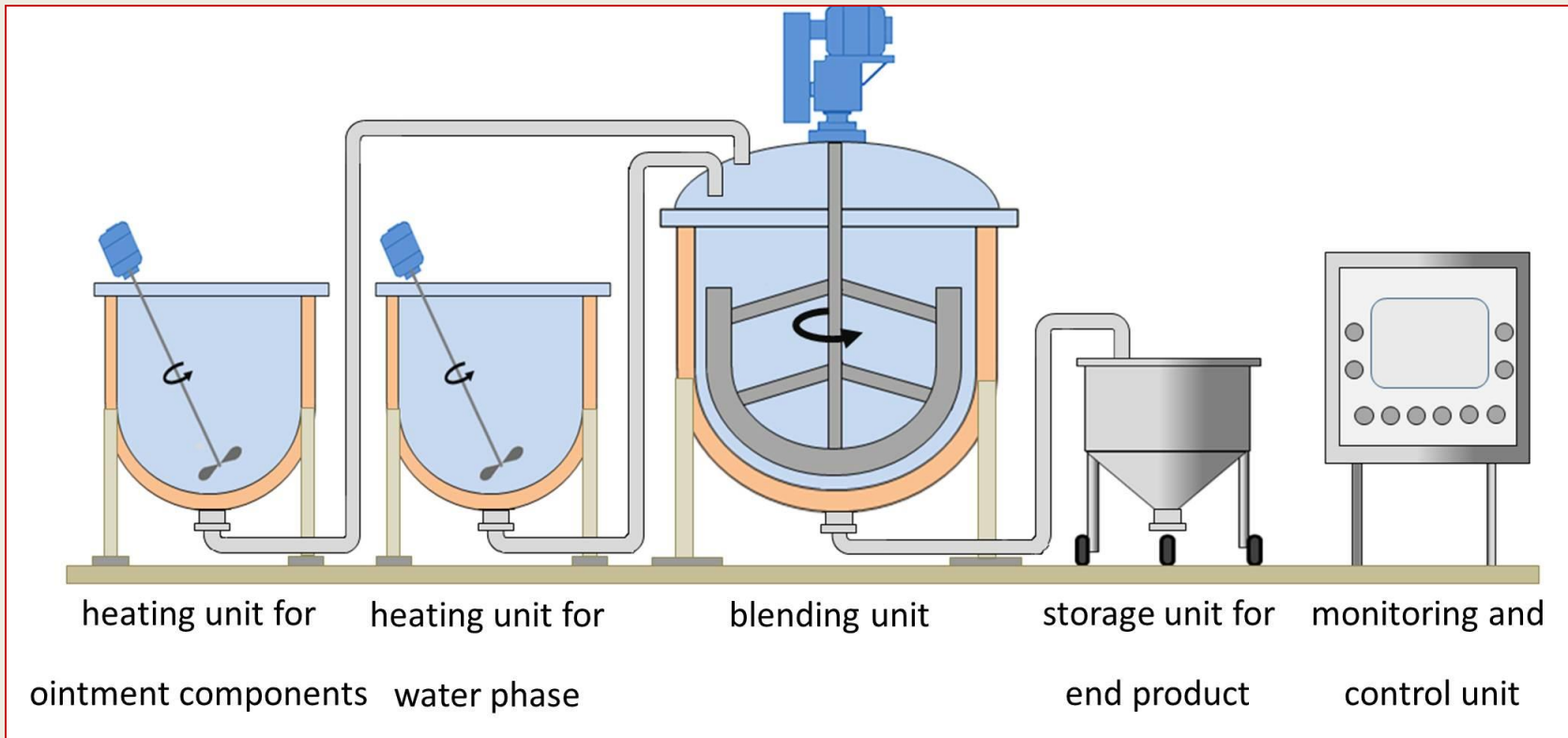
Mixing of semisolids

Triple-roll-mill



Mixing of melted semisolids

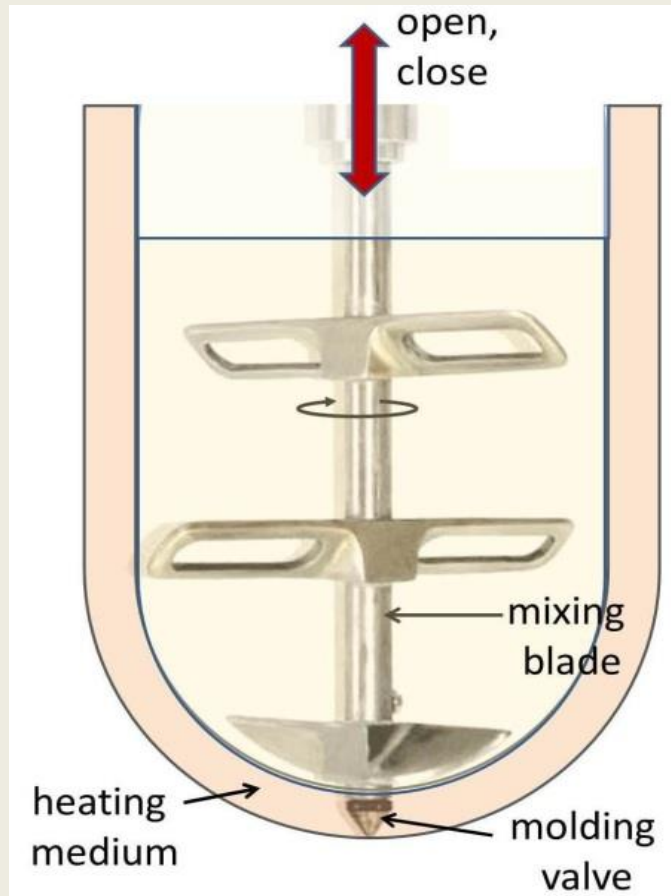
Preparation of cosmetics



Automated production line for making ointments and creams

Mixing of melted semisolids

Preparation of suppositories



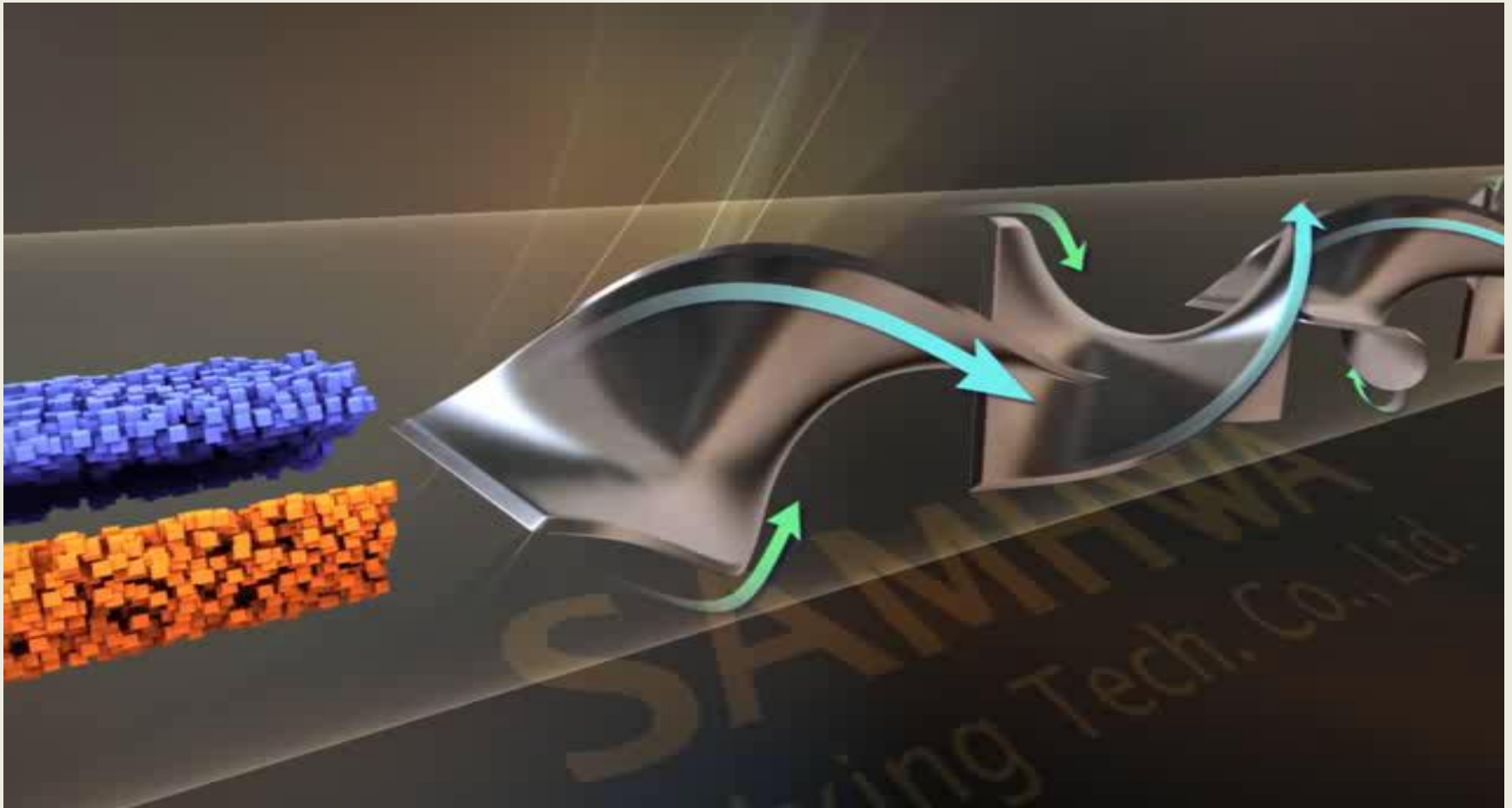
Mixing of melted semisolids

Preparation of suppositories



Mixing of melted semisolids

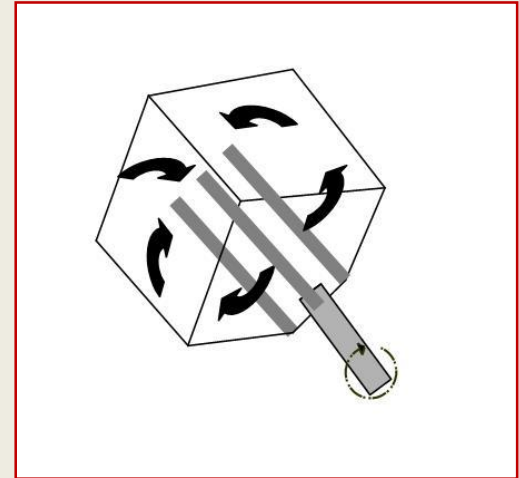
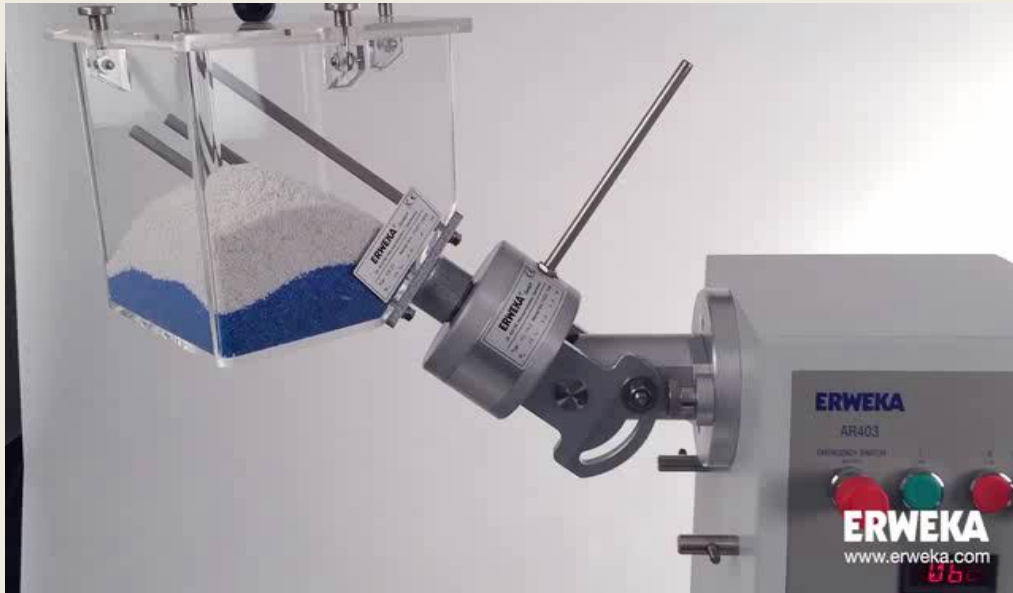
Static mixer



Mixing of solids

Mixing of solids

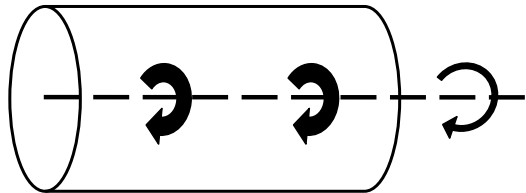
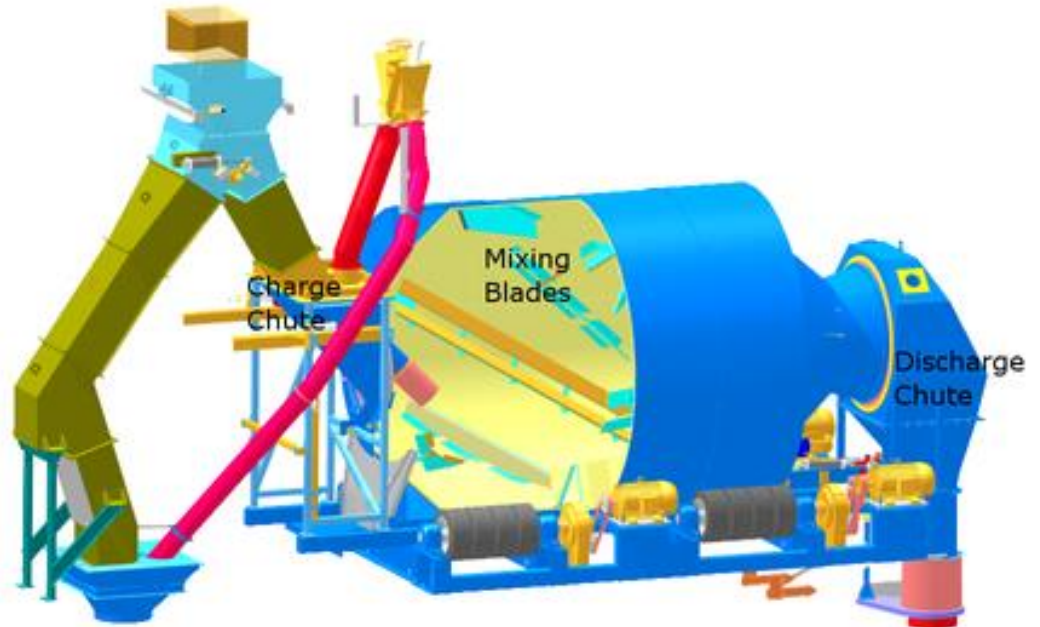
Rotating cube mixer



Mixing of solids

Drum mixer

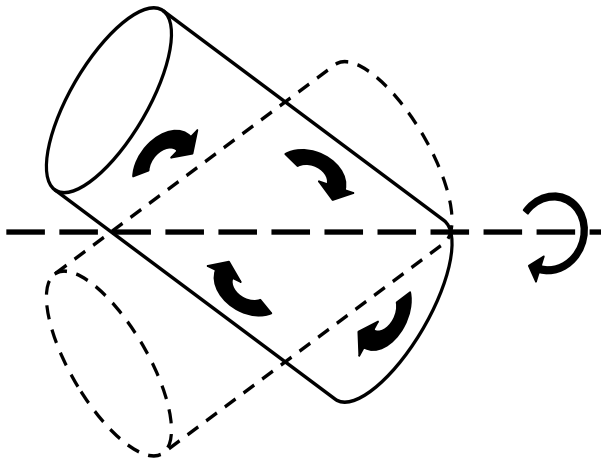
same direction of axis



Mixing of solids

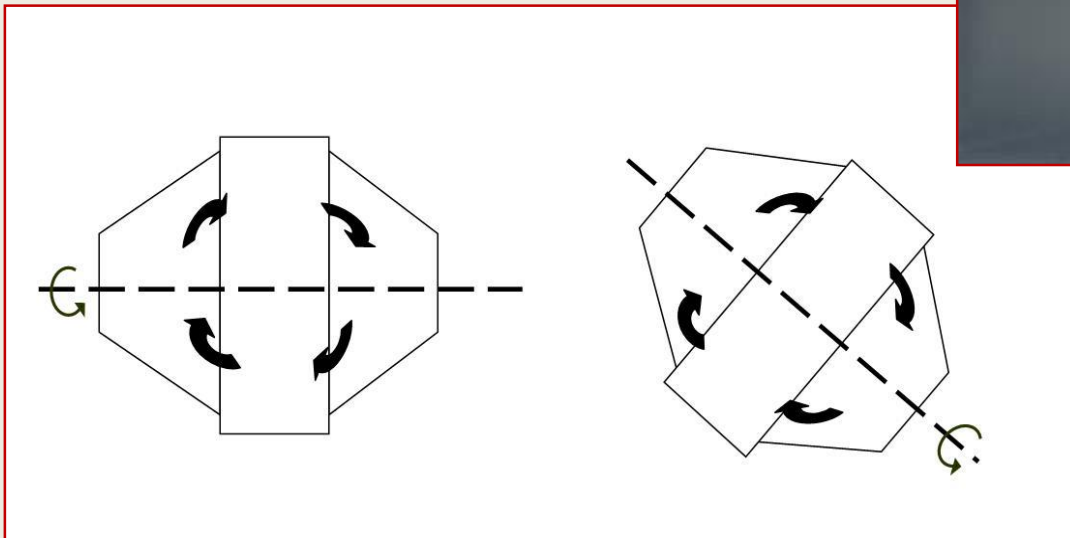
Drum mixer

alternate direction of axis



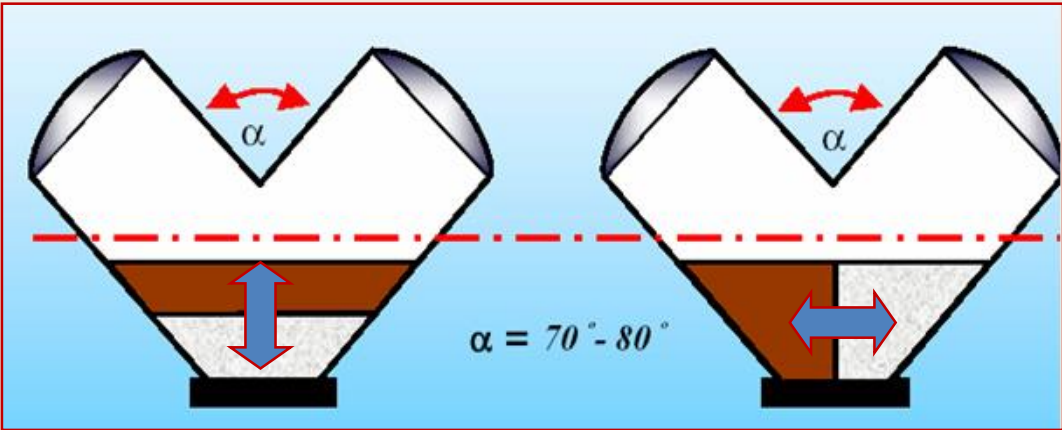
Mixing of solids

Double cone mixer



Mixing of solids

V-mixer

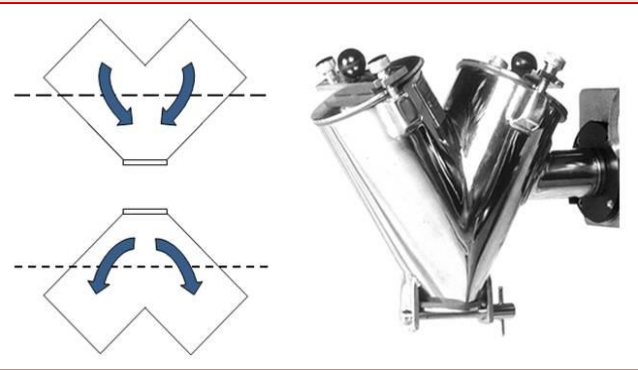


a./

The horizontal layers are mixed by each side.

b./

The vertical layers are mixed from one side to other.

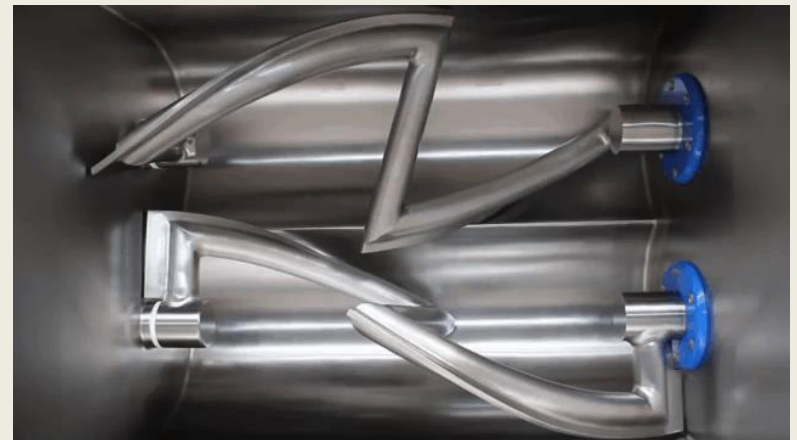
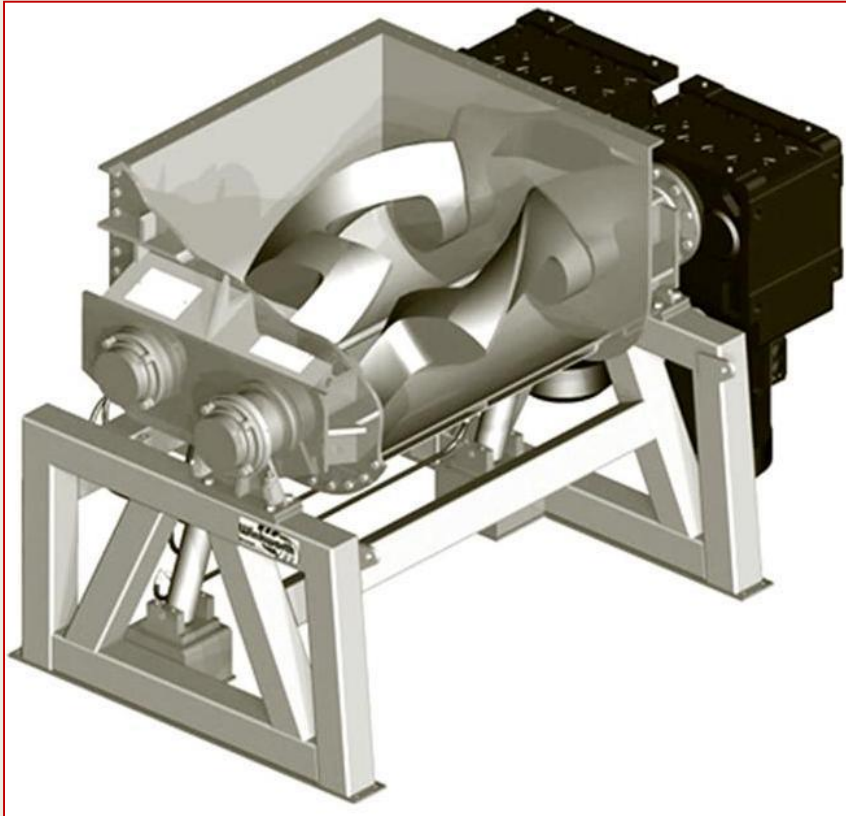


V-Blender
Top-Bottom Fill Side-to-Side Fill

Rotation Rate: 10 RPM
N. of Particles: 300,000
Particle Shape: Spherical
Playback Speed: 400% Faster

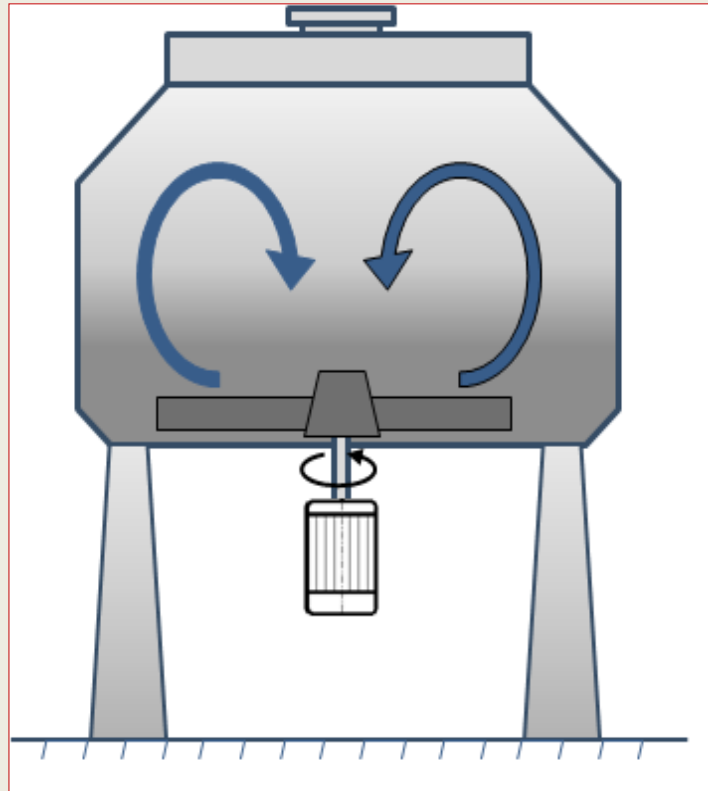
Mixing of solids

Z-arm mixer (Sigma blade mixer)



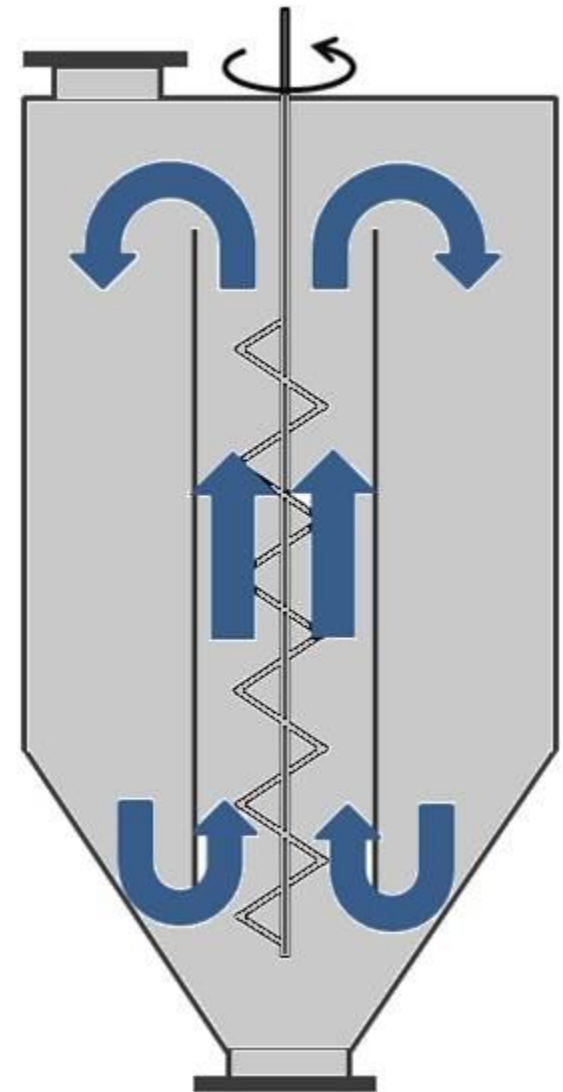
Mixing of solids

High shear mixer



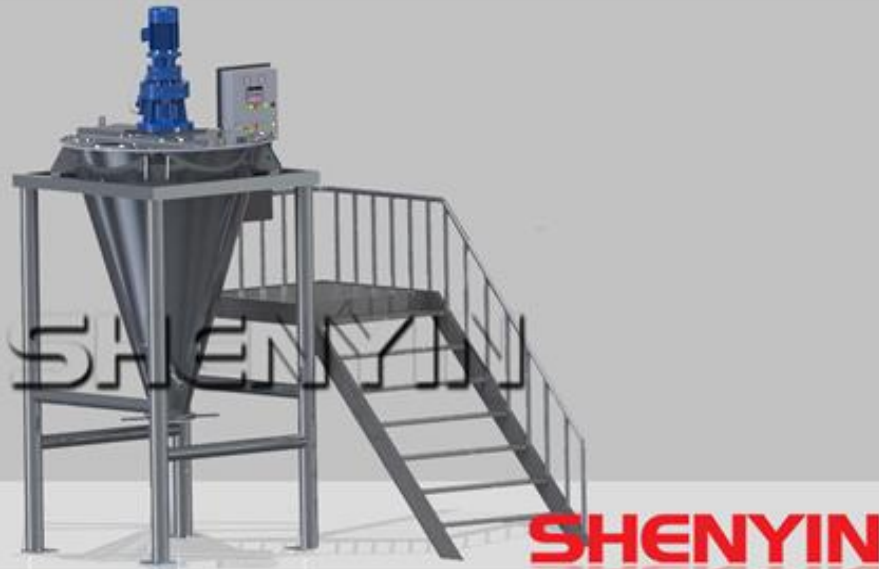
Mixing of solids

Vertical screw mixer



Mixing of solids

Conical vertical screw mixer



SHENYIN

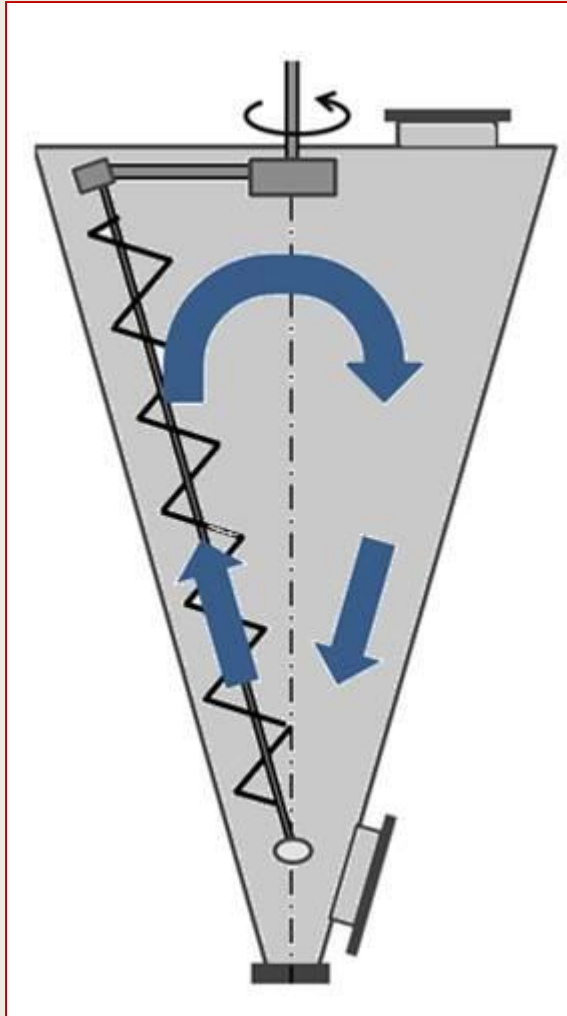
CE IAF COC SGS CNAS ISO 9001

Since 1983



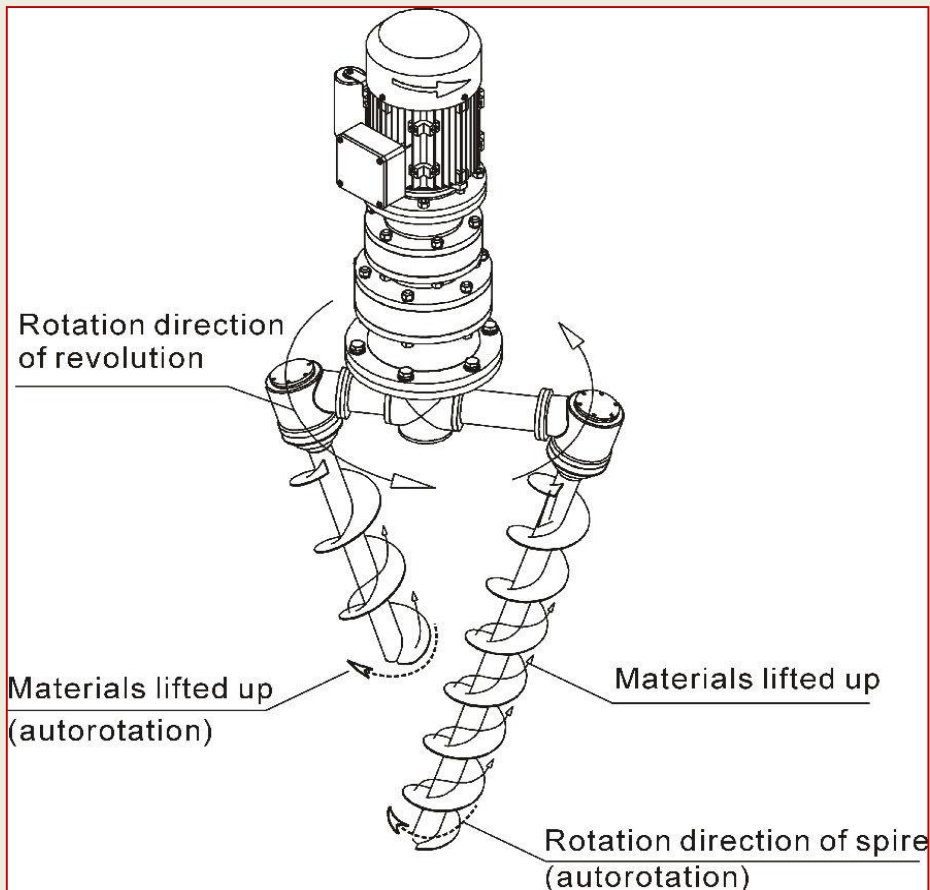
Mixing of solids

Conical oblique screw mixer



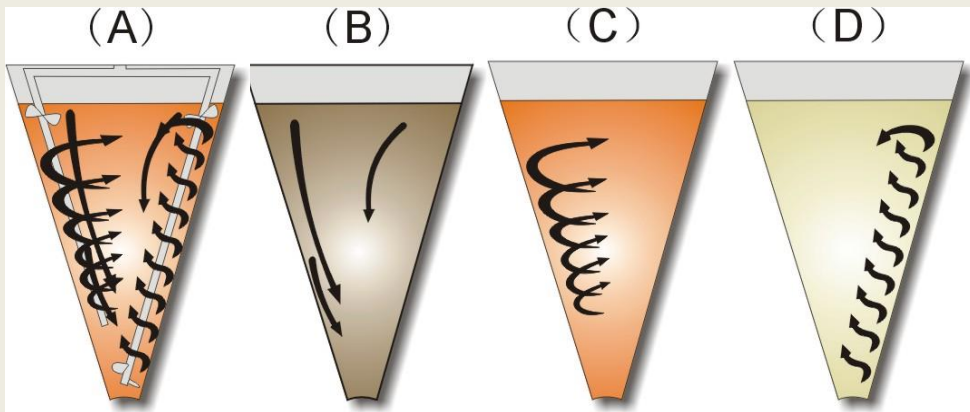
Mixing of solids

Conical oblique double screw mixer



Mixing of solids

Conical oblique double screw mixer



Mixing of solids

Drum mixer



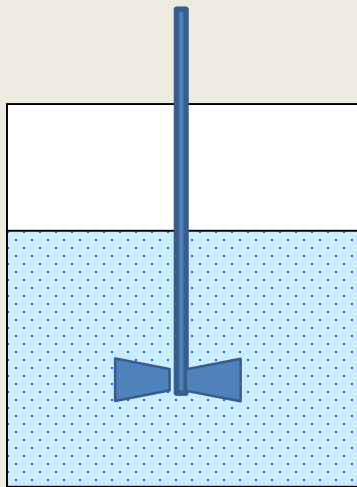
Controlling and optimization of mixing

Controlling and optimization of mixing

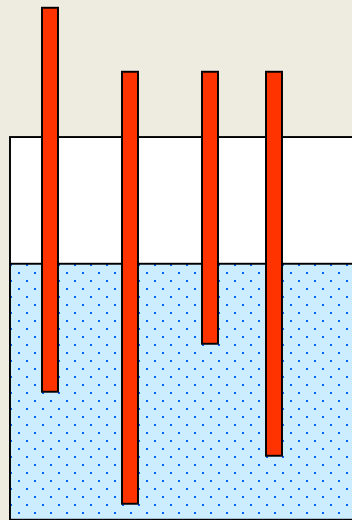
Examination of homogeneity

The proper number of samples are taken from different places of the system (liquid)

- determination of API content
- mean, standard deviation, boundary value analysis



mixing



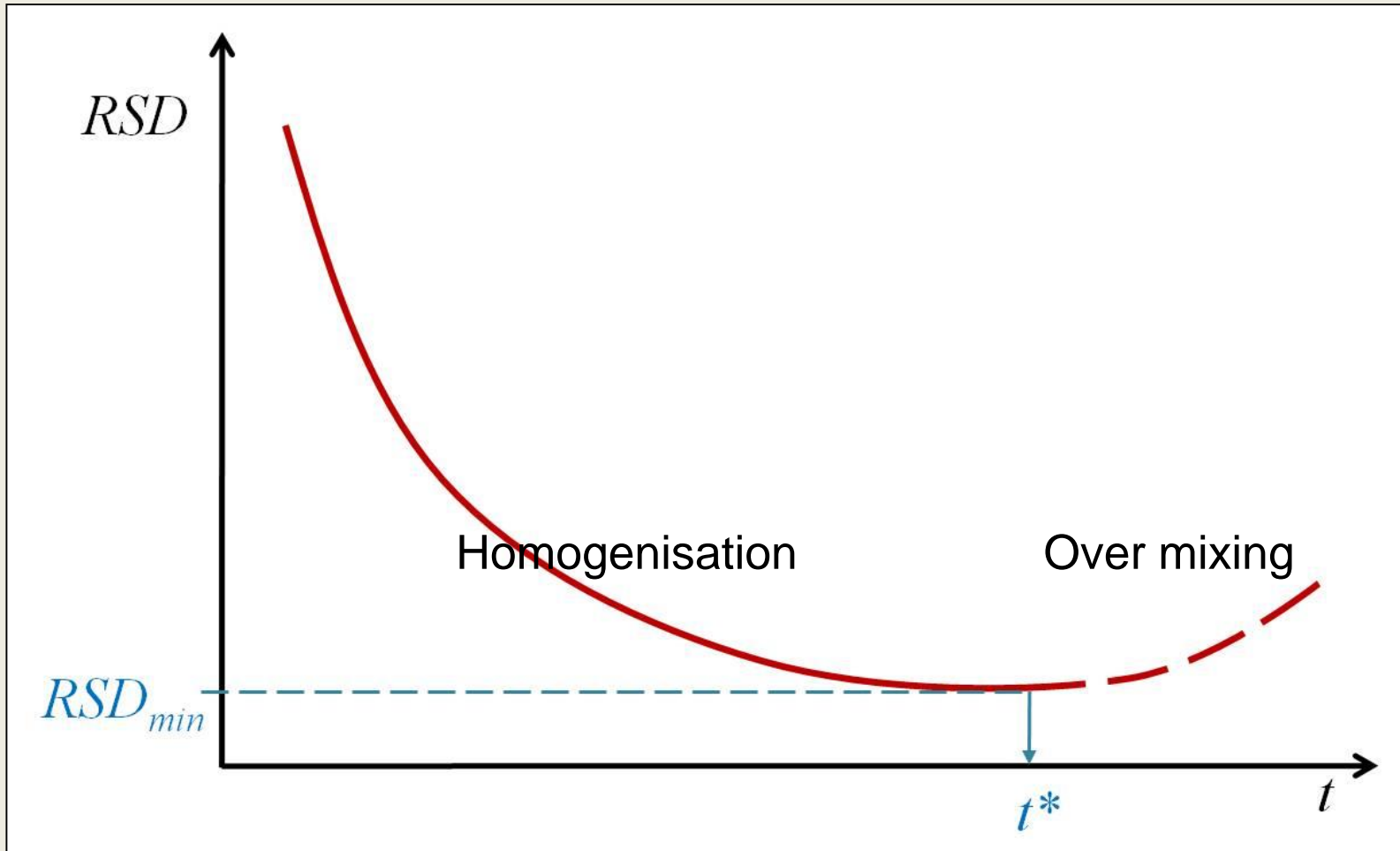
sampling



analysis

Controlling and optimization of mixing

Examination of homogeneity



***Thank you
for your attention!!!***